

Resources on Velikovsky's Ideas

Pensee 1973 Vol 3 Synthesis of Manna

http://www.catastrophism.com/intro/search.cgi?zoom_query=pensee&zoom_page=2&zoom_per_page=25

[http://www.catastrophism.com/intro/search.cgi?zoom_query=pensee+Vol+3&zoom_per_page=25&zoom_and=1&zoom_cat\[\]=-1](http://www.catastrophism.com/intro/search.cgi?zoom_query=pensee+Vol+3&zoom_per_page=25&zoom_and=1&zoom_cat[]=-1)

Architecture of the Earth RA Daley 1938

Sanders and Peck Hopi Indian Legends

<http://nativeamerican-art.com/hopi-legends.html>

Hydrocarbons in comet tail

http://www.windows2universe.org/headline_universe/astrophysics/hyakutake.html

Formaldehyde as a carbohydrate

https://ca.answers.yahoo.com/question/index;_ylt=A0LEVilL51VWIRUA02NjmolQ;_ylu=X3oDMTByb2h1ajk3BHNIYwNzcGRwb3MDOARjb2xvA2JmMQR2dGIkAw--?qid=20100727115841AAHFr1n

Best Answer: With the formula of CH₂O, it would seem to fit the definition of a carbohydrate. I checked some data on carbohydrates and came up with: Formerly the name "carbohydrate" was used in chemistry for any compound with the formula C_m(H₂O)_n. Following this definition, some chemists considered formaldehyde CH₂O to be the simplest carbohydrate, while others claimed that title for glycolaldehyde. Today the term is generally understood in the biochemistry sense, which excludes compounds with only one or two carbons. So it would appear that formaldehyde is not classified as a carbohydrate.

[Trevor H](#) · 5 years ago

A carbohydrate has the formula C_x(H₂O)_y, the formula for formaldehyde (methanal) is CH₂O so yes it's the simplest carbohydrate you can form.

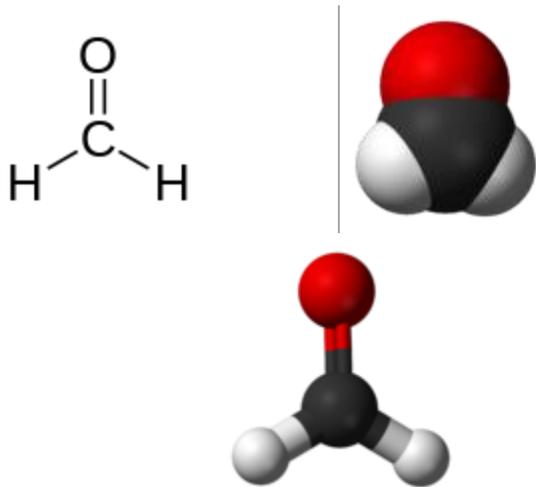
[Dancing Cactoid](#) · 5 years ago

Can it form in the atmosphere? <https://en.m.wikipedia.org/wiki/Formaldehyde>

Formaldehyde

[Methanal](#) redirects here. It is not to be confused with [methanol](#) or [menthol](#).
For other uses, see [Formaldehyde \(disambiguation\)](#).

Formaldehyde



Formaldehyde is a naturally-occurring [organic compound](#) with the [formula](#) CH₂O. It is the simplest [aldehyde](#) and is also known by its [systematic name](#) **methanal**. The common name of this substance comes from its similarity and relation to [formic acid](#).

Formaldehyde is an important precursor to many other materials and chemical compounds. In 1996, the installed capacity for the production of formaldehyde was estimated to be 8.7 million tons per year.^[7] It is mainly used in the production of industrial [resins](#), e.g., for particle board and coatings.

In view of its widespread use, toxicity, and volatility, formaldehyde is a significant consideration for human health.^[8] In 2011, the US [National Toxicology Program](#) described formaldehyde as "[known to be a human carcinogen](#)".^{[9][10][11]}